

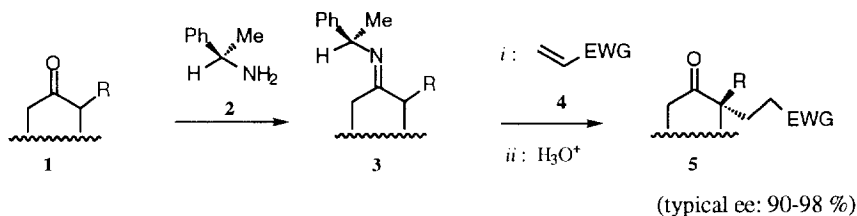
Asymmetric Bridging Annulation Reaction Involving the Intramolecular Conjugate Addition of Chiral Imines to Enoates: Access to a Polycyclic System Related to the Taxane Core

Christian Cavé, Sophie Boggero, Ramon Casas¹, Françoise Dumas, Jacqueline Mahuteau, and Jean d'Angelo*

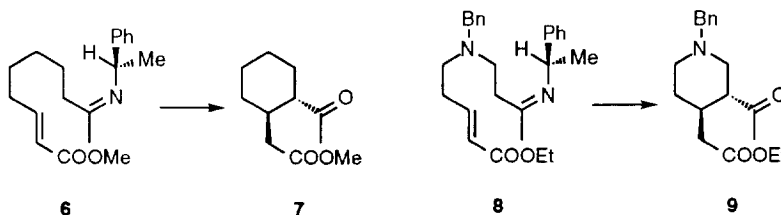
Unité de Chimie Organique Associée au CNRS, Centre d'Etudes Pharmaceutiques, Université Paris-Sud, 5, rue J.-B. Clément, 92296 Châtenay-Malabry (France)

Abstract : (*R*)-1-phenylethylamine-induced cyclization of ketoenoate **2** **4** led to a 2:1 mixture of "all-*cis*" polycyclic adducts **2** **5** and **2** **6**, structurally related to the taxane series.

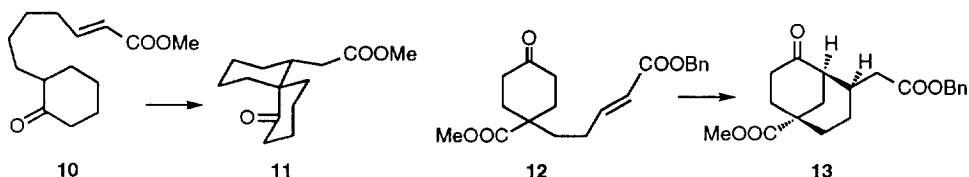
Ten years ago, we reported that chiral imines **3**, derived from *racemic* 2-alkylcyclohexanones **1** and optically active 1-phenylethylamine **2**, add, *under neutral conditions*, to electrophilic alkenes **4** leading, after hydrolytic work-up, to 2,2-dialkylcyclohexanones **5** with a good yield and excellent regio- and stereoselectivity.²



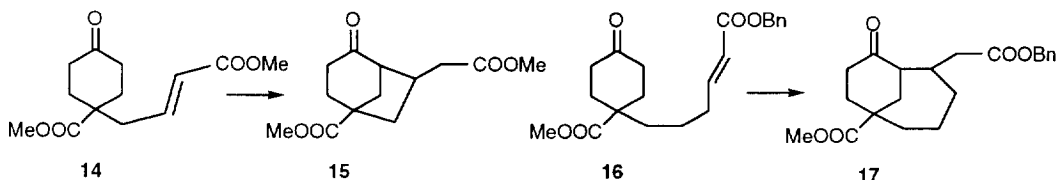
This Michael addition, which tolerates a great variation in the nature of the starting cyclohexanone and the electrophilic alkene, constitutes indisputably one of the most potent methodology for the *enantioselective elaboration of quaternary carbon centers*. In this respect, it has been successfully applied by ourselves and others to the asymmetric synthesis of various compounds of natural origin, including terpenes, steroids and alkaloids.² The mechanistic aspects of this reaction have also been extensively investigated, from both experimental and theoretical viewpoints.² Several *intramolecular variants* of this Michael addition have been developed; thus, for example, chiral imines **6**³ and **8**⁴ underwent a facile *carbocyclization* under thermal conditions, furnishing, after hydrolytic work-up, derivatives **7** and **9**, respectively, with an excellent enantioselectivity (*ca* 90%).



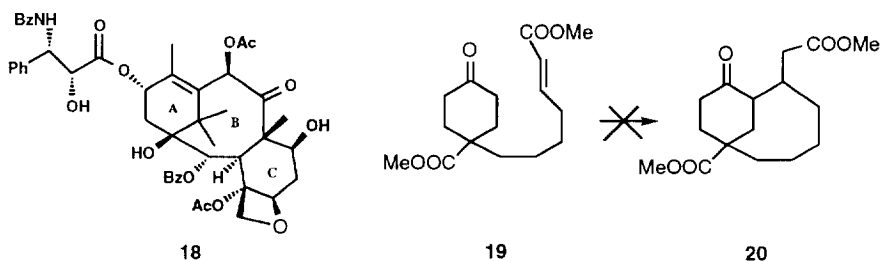
On the other hand, (*R*)-1-phenylethylamine-induced *spiroannulation* of ketoenone **10** led to ketoester (**1S**, **2R**)-**11** with a high degree of control ($\geq 90\%$) of the two newly created stereogenic centers.⁵ In a similar fashion, *bridging annulation* of ketoenone **12**, induced by (*ent*)-**2**, gave the bicyclo[3.3.1]nonane derivative **13** with a high level of stereoselectivity (100% de, 90% ee).⁶



However, this bridging annulation proved to be very sensitive to the size of the newly created ring; thus, the [3.2.1] and [4.3.1] bicyclic compounds **15** and **17**, elaborated from ketoenones **14** and **16**, respectively, were both obtained with a low degree of stereoselectivity (60% de, 30% ee and 20% de, respectively).⁶

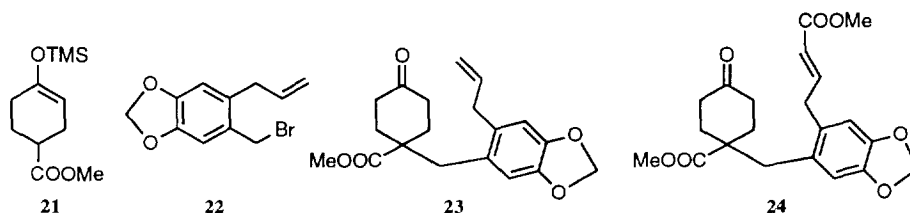


In connection with our synthetic efforts in the taxane series (exemplified by taxol **18**)⁷, we recently planned the construction of the [AB] ring framework of these systems, by extending the present bridging annulation. However, all attempts at elaborating the bicyclo[5.3.1]undecane derivative **20** by 1-phenylethylamine-induced cyclization of ketoenone **19** were invariably unsuccessful, a reflect of the difficulties usually encountered in establishing eight-membered carbocycles from open-chain precursors.⁶

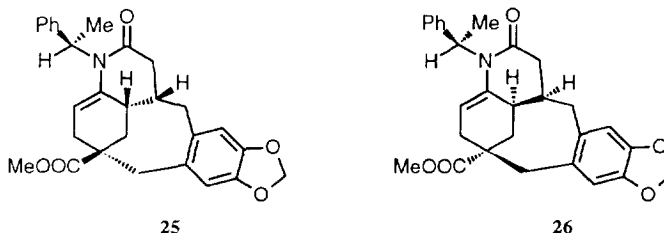


In light of the preceding observation, we reasoned that the intercalation of an aromatic moiety, precursor of the future C ring of taxanes, in the side-chain of **19**, by reducing significantly the degrees of freedom of the system, as well as the transannular strain in the eight-membered adduct, would favor the annulation reaction.⁸ This was tested on model ketoenone **24**, efficiently elaborated by condensing first the lithio derivative of ester **21**⁹ with bromomethylsafrole **22**¹⁰ (*i* : **21**, 1.5 eq of LDA, THF, 30 min at -78°C ; *ii* : **22**, $-78^\circ\text{C} \rightarrow 20^\circ\text{C}$, then 12 h at this temperature; *iii* : 1N HCl, 1 h at 20°C). Compound **23**, thus obtained with a 58% overall yield, was

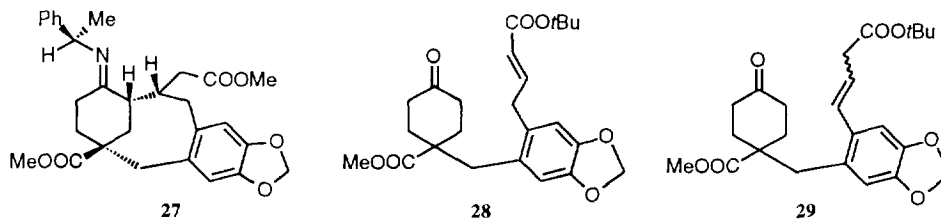
then converted into ketoenolate **24**¹¹ through a two-step sequence (*i* : *cat* OsO₄, NaIO₄, Et₂O, H₂O, 1 h at 20 °C; *ii* : Ph₃P=CH-COOMe, CH₂Cl₂, 24 h at 20 °C, 70 % overall yield).



Ketoenolate **24** was then cyclized [2 eq of (*R*)-1-phenylethylamine, 72 h in refluxing toluene, with removal of water], leading with a 48 % yield to a 2:1 mixture of isomeric adducts **25**¹² and **26**¹³, respectively. These were separated by preparative HPLC¹⁴ and fully characterized. The same "all-*cis*" relative stereochemistry in the tricyclic systems of **25** and **26** was established through the complete assignments of the ¹H and ¹³C NMR resonances, including 1D and 2D experiments, PFG-phase sensitive DQF COSY, and ¹H-¹³C direct and long range correlations (PFG-HMQC and PFG-HMBC). The proposed absolute configurations of the three newly created stereogenic centers in **25** and **26**, although not definitively established, rests on the mechanism of the present Michael addition, assuming that the addition took place predominantly on the less hindered side of the intermediary chiral imine, *anti* to the phenyl group of the inducer.^{2,6}



It is clear that the formation of adducts **25** and **26** from **24** involves the intermediary Michael adducts (*e.g.* **27**) which underwent a lactamization side-reaction. In order to minimize such a lactamization process, 1-phenylethylamine-induced cyclization of ketoenolate **28**, the *tert*-butyl ester analog of **24**, was next investigated. However, when **28** was submitted to the aforementioned cyclization conditions, only the β,γ -ethylenic ester **29** was isolated.



An efficient access of a polycyclic system related to the taxane core has thus been achieved. New synthetic developments of this methodology are currently under investigation in our laboratory.

NOTES AND REFERENCES

- 1 EC Postdoctoral Fellow, on leave from the Universtiat Autonoma de Barcelona (Spain).
- 2 Reviews: d'Angelo, J.; Desmaële, D.; Dumas, F.; Guingant, A. *Tetrahedron: Asymmetry*, **1992**, *3*, 459-505. d'Angelo, J.; Cavé, C.; Desmaële, D.; Dumas, F. *Trends in Organic Synthesis*, Pandalalai, S.G., Ed.; Trivandrum (India), **1993**, *4*, 555-616.
- 3 Dumas, F.; d'Angelo, J. *Tetrahedron: Asymmetry*, **1990**, *1*, 167-170.
- 4 Hirai, Y.; Terada, T.; Yamazaki, T. *J. Amer. Chem. Soc.*, **1988**, *110*, 958-960.
- 5 d'Angelo, J.; Ferroud, C. *Tetrahedron Lett.*, **1989**, *30*, 6511-6514.
- 6 Dumas, F.; Maine, V.; Cavé, C.; d'Angelo, J. *Tetrahedron: Asymmetry*, **1994**, *5*, 339-342.
- 7 Review: Nicolaou, K. C.; Dai, W.-M., Guy, R. K. *Angew. Chem. Int. Ed. Engl.*, **1994**, *33*, 15-44. Holton, R. A.; Somoza, C.; Kim, H.-B.; Liang, F.; Biediger, R. J.; Boatman, P. D.; Shindo, M.; Smith, C. C.; Kim, S.; Nadizadeh, H.; Suzuki, Y.; Tao, C.; Vu, P.; Tang, S.; Zhang, P.; Murthi, K. K.; Gentile, L. N.; Liu, J. H. *J. Amer. Chem. Soc.*, **1994**, *116*, 1597-1600. Nicolaou, K. C.; Ueno, H.; Liu, J. J.; Nanterm, P. G.; Yang, Z.; Renaud, J.; Paulvannan, K.; Chadha, R.J. *Amer. Chem. Soc.*, **1995**, *117*, 624-659 and references cited therein.
- 8 Horiguchi, Y.; Furukawa, T.; Kuwajima, I. *J. Amer. Chem. Soc.*, **1989**, *111*, 8277-8279.
- 9 Jung, M. E.; Mc Combs, C. A. *Tetrahedron Lett.*, **1976**, 2935-2938.
- 10 Bromomethylsafrole **22** was prepared with a 86% yield by refluxing 2 h a solution of NaBr in acetone with chloromethylsafrole (Lurik, B. B.; Volkov, Y. P. *Zh. Org. Khim.*, **1986**, *22*, 384-387).
- 11 **24**: oil; ^1H NMR (200 MHz, CDCl_3) δ 6.93 (dt, $J = 15.2$ Hz, $J = 6.1$ Hz, 1H) 6.50 (s, 1H) 6.45 (s, 1H) 5.86(s, 2H) 5.58(dt, $J = 15.2$ Hz, $J = 3.0$ Hz, 1H) 3.67 (s, 3H) 3.64 (s, 3H) 3.35 (dd, $J = 6.1$ Hz, $J = 3.0$ Hz, 1H) 2.75 (s, 2H) 2.5-2.2 (m, 6H) 1.7-1.6 (m, 2H); ^{13}C NMR (50 MHz, CDCl_3) δ 210.4 (C), 177.5 (2 C), 147.2 (C), 146.9 (CH), 129.6 (C), 127.6 (2 C), 122.0 (CH), 110.7 (CH), 110.0 (CH), 101.1 (CH₂), 56.2 (CH₃), 52.1(CH₃), 48.4 (C), 41.7 (CH₂), 38.5 (2 CH₂), 35.4 (CH₂), 33.7 (2 CH₂).
- 12 **25**: solid, mp 210-212 °C (MeOH); $[\alpha]_{\text{D}}^{20} + 49.1$ (c 1.7, CCl_4); MS (70 eV): 459 (M⁺), 355, 296, 206, 175, 149, 105; IR (CHCl₃): 1729, 1660, 1633 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.35-7.20 (m, 5H), 6.65 (s, 1H), 6.43 (q, 1H, $J = 7.1$ Hz), 6.40 (s, 1H), 5.90 (s, 2H), 4.99 (dt, 1H, $J = 7.0$ Hz, $J = 2.0$ Hz), 3.69 (s, 3H), 3.04 (dd, 1H, $J = 13.5$ Hz, $J = 13.2$ Hz), 2.95 (d, 1H, $J = 14.5$ Hz), 2.92 (dd, 1H, $J = 17.9$ Hz, $J = 7.0$ Hz), 2.83 (d, 1H, $J = 14.6$ Hz), 2.75 (ddd, 1H, $J = 10.0$ Hz, $J = 2.0$ Hz, $J = 2.0$ Hz), 2.66 (d, 1H, $J = 17.9$ Hz), 2.44 (d, 1H, $J = 13.5$ Hz), 2.30 (ddd, 1H, $J = 16.5$ Hz, $J = 2.0$ Hz, $J = 2.0$ Hz), 2.15 (ddd, 1H, $J = 16.5$ Hz, $J = 7.0$ Hz, $J = 3.0$ Hz), 2.04 (dd, 1H, $J = 14.1$ Hz, $J = 10.0$ Hz), 2.00 (dd, 1H, $J = 13.2$ Hz, $J = 7.0$ Hz), 1.71 (d, 3H, $J = 7.1$ Hz), 1.50 (dd, 1H, $J = 14.9$ Hz, $J = 3.0$ Hz); ^{13}C NMR (50 MHz, CDCl_3) δ 176.0 (C), 168.5 (C), 146.7 (2 C), 141.8 (C), 134.2 (C), 133.4 (C), 129.9 (C), 128.7 (2 CH), 126.5 (CH), 125.4 (2 CH), 110.0 (CH), 109.7 (CH), 107.7 (CH), 100.9 (CH₂), 52.1 (CH₃), 49.9 (CH), 45.7 (C), 42.8 (CH₂), 39.5 (CH), 37.8 (CH₂), 37.5 (CH₂), 36.1 (CH), 33.8 (CH₂), 30.0 (CH₂), 15.5 (CH₃).
- 13 **26**: amorphous solid; $[\alpha]_{\text{D}}^{20} + 54.0$ (c 3.6, CCl_4); MS (70 eV): 459 (M⁺), 355, 296, 206, 175, 149, 105; IR (CHCl₃): 1729, 1660, 1633 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.35-7.20 (m, 5H), 6.65 (s, 1H), 6.42 (s, 1H), 6.10 (q, 1H, $J = 7.0$ Hz), 5.90 (s, 2H), 5.04 (ddd, 1H, $J = 5.0$ Hz, $J = 2.5$ Hz, $J = 2.0$ Hz), 3.68 (s, 3H), 3.16 (d, 1H, $J = 14.0$ Hz), 3.00 (dd, 1H, $J = 17.0$ Hz, $J = 7.0$ Hz), 2.97 (d, 1H, $J = 14.0$ Hz), 2.94 (dd, 1H, $J = 14.0$ Hz, $J = 6.0$ Hz), 2.74 (dd, 1H, $J = 2.5$ Hz, $J = 2.0$ Hz), 2.61 (d, 1H, $J = 17.0$ Hz), 2.38 (d, 1H, $J = 14.0$ Hz), 2.24 (ddd, 1H, $J = 17.0$ Hz, $J = 5.0$ Hz, $J = 3.0$ Hz), 2.12 (ddd, 1H, $J = 17.0$ Hz, $J = 2.5$ Hz, $J = 2.5$ Hz), 2.02-1.95 (m, 2H), 1.82 (d, 3H, $J = 7.0$ Hz), 1.50 (dd, 1H, $J = 15.0$ Hz, $J = 3.0$ Hz); ^{13}C NMR (50 MHz, CDCl_3) δ 176.5 (C), 167.6 (C), 146.0 (2 C), 141.0 (C), 134.2 (C), 133.8 (C), 128.8 (C), 128.2 (2 CH), 126.2 (CH), 125.7 (2 CH), 109.8 (CH), 109.3 (CH), 106.7 (CH), 100.0 (CH₂), 51.8 (CH₃), 51.7 (CH), 45.0 (C), 42.7 (CH₂), 39.1 (CH), 37.4 (CH₂), 36.9 (CH₂), 36.0 (CH), 33.6 (CH₂), 29.7 (CH₂), 17.2 (CH₃).
- 14 Analytical HPLC: Hichrom column (Spherisorb S 5 W), length: 25 cm, internal diameter: 4.9 mm; detection: UV at 254 nm; eluent: cyclohexane/AcOEt 90:10; flow rate: 2 ml min⁻¹; retention times: **25**: 7.79 min, **26**: 6.83 min.

Acknowledgment. We thank Dr A. Commerçon (Société Rhône-Poulenc Rorer) for stimulating discussions, Dr A. Talab (Société MERAM, Melun, France) for GC-MS analyses, and Mr E. Rolim de Oliveira for an improvement in the synthesis of chloromethylsafrole.

This work has been supported by a Grant from the European Communities (The Rational Design of New Organic Molecules and Synthetic Methods, Contract Nr ERBCHRXCT 930141).